

# Physical Chemistry, Spectroscopy

Programme course

6 credits

Fysikalisk kemi, spektroskopi

NKEB09

Valid from: 2019 Spring semester

**Determined by**

Board of Studies for Chemistry, Biology  
and Biotechnology

**Date determined**

2018-08-31

## Main field of study

Chemistry

## Course level

First cycle

## Advancement level

G1X

## Course offered for

- Chemical Biology, Bachelor's Programme
- Chemistry - Molecular Design, Bachelor's Programme

## Entry requirements

Note: Admission requirements for non-programme students usually also include admission requirements for the programme and threshold requirements for progression within the programme, or corresponding.

## Prerequisites

General Chemistry, Organic Chemistry, Biochemistry, Physical Chemistry, Thermodynamics, Mathematics

## Intended learning outcomes

The course aims to confer knowledge in physical chemistry within the areas kinetics, quantum mechanics and molecular spectroscopy and skills in problem solving and laboratory tasks within the areas listed in the course contents below. After successfully completed course the course participant will be able to:

- understand the basic principles underlying the physical processes covered in the course
- describe and discuss the physical phenomena as well as the kinetic and spectroscopic instruments and measurement techniques
- account for, explain and exemplify the theory
- perform calculations for and solve problems within the course content area.
- analyse and apply the acquired facts and skills on complex physical processes
- conclude whether obtained results are reasonable and reflect over fundamental physical-chemistry problems in science, nature and society
- work with physical-chemistry problems solitarily and in groups
- analyse, compile and account for the laboratory exercise results in laboratory reports.

## Course content

Lectures, lessons and laboratory exercises cover the following areas: Kinetics: rate equations and their integration, reaction order, half life, mechanisms, catalysis, photochemistry, techniques for measuring fast reactions, flow, flash and relaxation methods, activation energies and molecular reaction dynamics. Quantum mechanics: basic quantum-mechanical concepts, particle-in-a-box, harmonic and anharmonic oscillator, rotation on a ring and on a sphere, the hydrogen atom, plus elementary introduction to many-electron atomic and molecular electron structure. Spectroscopy: interaction between electromagnetic radiation and matter by absorption and emission, fluorescence and phosphorescence, lasers, rotational, vibrational and electron spectroscopic methods such as microwave, infrared, visible/UV, X-ray photoelectron and magnetic resonance spectroscopy.

## Teaching and working methods

The theory is presented during the lectures. Problem solving is practised during the lessons. In the laboratory exercises fundamental laboratory techniques and moments connected to the theory and problem solving are practised. The laboratory exercises, whose laboratory exercise instructions and safety rules must be studied prior to the exercise, are mandatory. Written reports are handed in within five workdays from the laboratory exercise occasion. For the laboratory exercises special premises are required.

## Examination

|      |                     |             |            |
|------|---------------------|-------------|------------|
| LAB1 | Laboratory work     | 1.5 credits | U, G       |
| TEN1 | Written examination | 4.5 credits | U, 3, 4, 5 |

The problems at the written examination probes to what extent the participant fulfils the aims of the course. For approved laboratory work carried-through laboratory exercises and approved reports are required. For grade 3, 4 and 5, respectively, on the course, grade 3, 4 and 5, respectively, on the written exam together with grade G on the laboratory work are required.

## Grades

Four-grade scale, LiU, U, 3, 4, 5

## Other information

Supplementary courses: Materials Chemistry, Chemical Bonding, Computational Chemistry, Nanochemistry A and B, Coordination Chemistry, Physical-Organic Chemistry, Biological Measurements.

## Department

Institutionen för fysik, kemi och biologi

## Director of Studies or equivalent

Magdalena Svensson

## Examiner

Lars Ojamäe

## Course website and other links

<http://www.ifm.liu.se/physchem/courses/>

## Education components

Preliminary scheduled hours: 56 h

Recommended self-study hours: 104 h

## Course literature

P. Atkins & J. de Paula: Atkins' Physical Chemistry Laborationskompndier, problemsamling och PM för laborationsredogörelser.

## Common rules

### Course syllabus

A syllabus has been established for each course. The syllabus specifies the aim and contents of the course, and the prior knowledge that a student must have in order to be able to benefit from the course.

### Timetabling

Courses are timetabled after a decision has been made for this course concerning its assignment to a timetable module. A central timetable is not drawn up for courses with fewer than five participants. Most project courses do not have a central timetable.

### Interrupting a course

The vice-chancellor's decision concerning regulations for registration, deregistration and reporting results (Dnr LiU-2015-01241) states that interruptions in study are to be recorded in Ladok. Thus, all students who do not participate in a course for which they have registered must record the interruption, such that the registration on the course can be removed. Deregistration from a course is carried out using a web-based form: [www.lith.liu.se/for-studenter/kurskomplettering?l=sv](http://www.lith.liu.se/for-studenter/kurskomplettering?l=sv).

### Cancelled courses

Courses with few participants (fewer than 10) may be cancelled or organised in a manner that differs from that stated in the course syllabus. The board of studies is to deliberate and decide whether a course is to be cancelled or changed from the course syllabus.

### Regulations relating to examinations and examiners

Details are given in a decision in the university's rule book:  
<http://styrdokument.liu.se/Regelsamling/VisaBeslut/622678>.

### Forms of examination

#### Examination

Written and oral examinations are held at least three times a year: once immediately after the end of the course, once in August, and once (usually) in one of the re-examination periods. Examinations held at other times are to follow a decision of the board of studies.

Principles for examination scheduling for courses that follow the study periods:

- courses given in VT1 are examined for the first time in March, with re-

examination in June and August

- courses given in VT2 are examined for the first time in May, with re-examination in August and October
- courses given in HT1 are examined for the first time in October, with re-examination in January and August
- courses given in HT2 are examined for the first time in January, with re-examination at Easter and in August.

The examination schedule is based on the structure of timetable modules, but there may be deviations from this, mainly in the case of courses that are studied and examined for several programmes and in lower grades (i.e. 1 and 2).

- Examinations for courses that the board of studies has decided are to be held in alternate years are held only three times during the year in which the course is given.
- Examinations for courses that are cancelled or rescheduled such that they are not given in one or several years are held three times during the year that immediately follows the course, with examination scheduling that corresponds to the scheduling that was in force before the course was cancelled or rescheduled.
- If teaching is no longer given for a course, three examination occurrences are held during the immediately subsequent year, while examinations are at the same time held for any replacement course that is given, or alternatively in association with other re-examination opportunities. Furthermore, an examination is held on one further occasion during the next subsequent year, unless the board of studies determines otherwise.
- If a course is given during several periods of the year (for programmes, or on different occasions for different programmes) the board or boards of studies determine together the scheduling and frequency of re-examination occasions.

### **Registration for examination**

In order to take an examination, a student must register in advance at the Student Portal during the registration period, which opens 30 days before the date of the examination and closes 10 days before it. Candidates are informed of the location of the examination by email, four days in advance. Students who have not registered for an examination run the risk of being refused admittance to the examination, if space is not available.

Symbols used in the examination registration system:

\*\* denotes that the examination is being given for the penultimate time.

\* denotes that the examination is being given for the last time.

### **Code of conduct for students during examinations**

Details are given in a decision in the university's rule book:  
<http://styrdokument.liu.se/Regelsamling/VisaBeslut/622682>.

### **Retakes for higher grade**

Students at the Institute of Technology at LiU have the right to retake written examinations and computer-based examinations in an attempt to achieve a higher grade. This is valid for all examination components with code "TEN" and "DAT". The same right may not be exercised for other examination components, unless otherwise specified in the course syllabus.

### **Retakes of other forms of examination**

Regulations concerning retakes of other forms of examination than written examinations and computer-based examinations are given in the LiU regulations for examinations and examiners,

<http://stydokument.liu.se/Regelsamling/VisaBeslut/622678>.

### **Plagiarism**

For examinations that involve the writing of reports, in cases in which it can be assumed that the student has had access to other sources (such as during project work, writing essays, etc.), the material submitted must be prepared in accordance with principles for acceptable practice when referring to sources (references or quotations for which the source is specified) when the text, images, ideas, data, etc. of other people are used. It is also to be made clear whether the author has reused his or her own text, images, ideas, data, etc. from previous examinations.

A failure to specify such sources may be regarded as attempted deception during examination.

### **Attempts to cheat**

In the event of a suspected attempt by a student to cheat during an examination, or when study performance is to be assessed as specified in Chapter 10 of the Higher Education Ordinance, the examiner is to report this to the disciplinary board of the university. Possible consequences for the student are suspension from study and a formal warning. More information is available at <https://www.student.liu.se/studenttjanster/lagar-regler-rattigheter?l=sv>.

### **Grades**

The grades that are preferably to be used are Fail (U), Pass (3), Pass not without distinction (4) and Pass with distinction (5). Courses under the auspices of the faculty board of the Faculty of Science and Engineering (Institute of Technology) are to be given special attention in this regard.

1. Grades U, 3, 4, 5 are to be awarded for courses that have written examinations.
2. Grades Fail (U) and Pass (G) may be awarded for courses with a large degree of practical components such as laboratory work, project work and group work.

### **Examination components**

1. Grades U, 3, 4, 5 are to be awarded for written examinations (TEN).
2. Grades Fail (U) and Pass (G) are to be used for undergraduate projects and other independent work.

3. Examination components for which the grades Fail (U) and Pass (G) may be awarded are laboratory work (LAB), project work (PRA), preparatory written examination (KTR), oral examination (MUN), computer-based examination (DAT), home assignment (HEM), and assignment (UPG).
4. Students receive grades either Fail (U) or Pass (G) for other examination components in which the examination criteria are satisfied principally through active attendance such as other examination (ANN), tutorial group (BAS) or examination item (MOM).

The examination results for a student are reported at the relevant department.

### **Regulations (apply to LiU in its entirety)**

The university is a government agency whose operations are regulated by legislation and ordinances, which include the Higher Education Act and the Higher Education Ordinance. In addition to legislation and ordinances, operations are subject to several policy documents. The Linköping University rule book collects currently valid decisions of a regulatory nature taken by the university board, the vice-chancellor and faculty/department boards.

LiU's rule book for education at first-cycle and second-cycle levels is available at [http://styrdokument.liu.se/Regelsamling/Innehall/Utbildning\\_pa\\_grund-\\_och\\_avancerad\\_niva](http://styrdokument.liu.se/Regelsamling/Innehall/Utbildning_pa_grund-_och_avancerad_niva).