

## Chemistry 1 for Foundation Year

Kemi 1 för basår  
8 fup

Academic preparatory

BKL101

Valid from: 2026 Spring semester

<b>Determined by</b>	<b>Main field of study</b>	
Board of Studies for Chemistry, Biology and Biotechnology	No main field of study	
<b>Date determined</b>	<b>Course level</b>	<b>Progressive specialisation</b>
2025-08-28	Academic preparatory	
<b>Revised by</b>	<b>Disciplinary domain</b>	
	Natural sciences	
<b>Revision date</b>	<b>Subject group</b>	
	Chemistry	
<b>Offered first time</b>	<b>Offered for the last time</b>	
Autumn semester 2014		
<b>Department</b>	<b>Replaced by</b>	
Institutionen för fysik, kemi och biologi		

## Course offered for

- Foundation Year in Science and Technology

## Prerequisites

See admission requirements for the Foundation Year at Linköpings universitet.

## Intended learning outcomes

The aim of the course is to give students a foundation in chemistry and preparation for further university studies in technology and natural science. Completed course gives authority comparable with Chemistry 1. After completing the course, the student should:

- be able to plan and perform experimental chemical investigations according to safety regulations, be able to analyze and interpret the results and report these both orally and in writing through laboratory reports
- be able to explain the structure of atoms and ions, how elements are named and organized in the periodic table, and be able to name chemical compounds, interpret, write and use formulas for compounds and reactions
- be able to describe different types of chemical bonding based on the electron structure of the atoms and be able to relate chemical properties of a substance to the type and strength of the chemical bond and to the geometrical structure of the substance
- be able to write, balance and understand chemical equations and perform simple stoichiometric calculations
- be able to explain the concepts of pH, neutralization, strong and weak acids and bases, autoionization of water and buffers, as well as calculate pH values and results from acid-base titrations
- be able to use the concepts of entropy and enthalpy to discuss the driving force for a chemical reaction and be able to estimate the enthalpy change of a reaction, as well as be able to use the concepts of oxidation and reduction and describe electrochemical applications
- show basic knowledge of the structure, functional groups, nomenclature and basic physical properties of some classes of organic compounds, and explain the principles behind some methods for qualitative and quantitative chemical analysis

## Course content

**Introduction to the subject of chemistry:** What is chemistry? The nature and working methods of chemistry. Basic laboratory techniques and chemical safety. The role of chemistry in society and for sustainable development.

**The structure of the atom and the periodic table:** atomic model, chemical symbols, atomic mass, metals and nonmetals, groups and periods, electron distribution, excitation, isotope, mass number, valence electrons, noble gas configuration, ion.

**Chemical bonding and compounds:** elements and chemical compounds, chemical bonds and aggregation states, ionic compounds and molecular compounds, electronegativity, dipole, polarity, compound and hydrated ions, chemical name giving, melting and boiling points, single/double/triple bonds, water of crystallization

**Chemical reactions:** chemical reaction formulas, balancing of chemical reactions, equivalent quantities of substances, precipitation reactions, combustion reactions, decomposition reactions, gas phase reactions.

**Chemical calculations (stoichiometry):** amount of substance, molar mass, Avogadro's constant, accuracy and value figures, preparation of solutions and concentration of solutions, limiting reactant, excess reactant, yield, molar volume of gases.

**Acids and bases:** protolysis reactions, acid-base pairs, autoprotolysis of water, pH calculations, strong and weak acids, neutralization, pH indicator, buffer solutions.

**Thermochemistry:** energy conversion, enthalpy, entropy, exothermic and endothermic reactions, mean bond enthalpy, spontaneous reactions, activation energy, specific heat capacity, calorimetry.

**Oxidation and reduction:** electrochemical series of metals, complete and incomplete electron transfer, oxidation numbers, balancing redox reactions, electrochemistry, standard electrode potentials, galvanic cells, batteries, fuel cells, electrolysis and corrosion.

**Organic chemistry:** molecular formulas, structural formulas, functional groups, nomenclature, isomerism, substance classes, synthesis reactions, food substances, modern materials.

**Analytical chemistry:** an introduction to analytical chemistry, qualitative and quantitative analysis, titration, chromatography, reagents.

## Teaching and working methods

The course consists of lectures, lessons and laboratory work. The students homework, initiative and activity is of great importance. Laboratory work takes place in special laboratories.

## Examination

LAB1	Laboratory work	1.5 fup	U, G
TEN1	Written examination	6.5 fup	U, 3, 4, 5

Attempts to deceive by the use of prohibited aids or other methods during examinations or other forms of assessments of study performance may lead to a failed result on the examination/assessment in question.

Grades for examination modules are decided in accordance with the assessment criteria presented at the start of the course.

## Grades

Four-grade scale, LiU, U, 3, 4, 5

## Other information

Supplementary courses: Chemistry 2 for Foundation Year

### About teaching and examination language

The teaching language is presented in the Overview tab for each course. The examination language relates to the teaching language as follows:

- If teaching language is “Swedish”, the course as a whole could be given in Swedish, or partly in English. Examination language is Swedish, but parts of the examination can be in English.
- If teaching language is “English”, the course as a whole is taught in English. Examination language is English.
- If teaching language is “Swedish/English”, the course as a whole will be taught in English if students without prior knowledge of the Swedish language participate. Examination language is Swedish or English depending on teaching language.

### Other

The course is conducted in such a way that there are equal opportunities with regard to sex, transgender identity or expression, ethnicity, religion or other belief, disability, sexual orientation and age.

The planning and implementation of a course should correspond to the course syllabus. The course evaluation should therefore be conducted with the course syllabus as a starting point.

The course is campus-based at the location specified for the course, unless otherwise stated under “Teaching and working methods”. Please note, in a campus-based course occasional remote sessions could be included.

## Common rules

### Course syllabus

A syllabus must be established for each course. The syllabus specifies the aim and contents of the course, and the prior knowledge that a student must have in order to be able to benefit from the course.

### Timetabling

Courses are timetabled after a decision has been made for this course concerning its assignment to a timetable module.

### Interrupting a course and deregistration

The LiU decision, Guidelines concerning confirmation of participation in education, Dnr LiU-2020-02256 (<https://styrdokument.liu.se/Regelsamling/VisaBeslut/764582>), states that interruptions in study are to be recorded in Ladok. Thus, all students who do not participate in a course for which they have registered are therefore obliged to report the interruption so that this can be noted in Ladok. Deregistration from or interrupting a course is carried out using a [Web-based form](#).

### Cancelled courses and changes to the course syllabus

Courses with few participants (fewer than 10) may be cancelled or organised in a manner that differs from that stated in the course syllabus. The Dean is to deliberate and decide whether a course is to be cancelled or changed from the course syllabus.

### Guidelines relating to examinations and examiners

For details, see Guidelines for education and examination for first-cycle and second-cycle education at Linköping University, Dnr LiU-2023-00379 (<http://styrdokument.liu.se/Regelsamling/VisaBeslut/917592>).

An examiner must be employed as a teacher at LiU according to the LiU Regulations for Appointments, Dnr LiU-2022-04445 (<https://styrdokument.liu.se/Regelsamling/VisaBeslut/622784>). For courses in second-cycle, the following teachers can be appointed as examiner: Professor (including Adjunct and Visiting Professor), Associate Professor (including Adjunct), Senior Lecturer (including Adjunct and Visiting Senior Lecturer), Research Fellow, or Postdoc. For courses in first-cycle, Assistant Lecturer (including Adjunct and Visiting Assistant Lecturer) can also be appointed as examiner in addition to those listed for second-cycle courses. In exceptional cases, a Part-time Lecturer can also be appointed as an examiner at both first- and second cycle, see Delegation of authority for the Board of Faculty of Science and Engineering.

## Forms of examination

### Principles for examination

Written and oral examinations and digital and computer-based examinations are held at least three times a year: once immediately after the end of the course, once in August, and once (usually) in one of the re-examination periods. Examinations held at other times are to follow a decision of the board of studies.

Principles for examination scheduling for courses that follow the study periods:

- courses given in VT1 are examined for the first time in March, with re-examination in June and August
- courses given in VT2 are examined for the first time in May, with re-examination in August and October
- courses given in HT1 are examined for the first time in October, with re-examination in January and August
- courses given in HT2 are examined for the first time in January, with re-examination in March and in August.

The examination schedule is based on the structure of timetable modules, but there may be deviations from this, mainly in the case of courses that are studied and examined for several programmes and in lower grades (i.e. 1 and 2).

Examinations for courses that are cancelled or rescheduled such that they are not given in one or several years are held three times during the year that immediately follows the course, with examination scheduling that corresponds to the scheduling that was in force before the course was cancelled or rescheduled.

When a course, or a written or oral examination (TEN, DIT, DAT, MUN), is given for the last time, the regular examination and two re-examinations will be offered. Thereafter, examinations are phased out by offering three examinations during the following academic year at the same times as the examinations in any substitute course. The exception is courses given in the period HT1, where the three examination occasions are January, March and August. If there is no substitute course, three examinations will be offered during re-examination periods during the following academic year. Other examination times are decided by the board of studies. In all cases above, the examination is also offered one more time during the academic year after the following, unless the board of studies decides otherwise. In total, 6 re-examinations are offered, of which 2 are regular re-examinations. In the examination registration system, the examinations given for the penultimate time and the last time are denoted.

If a course is given during several periods of the year (for programmes, or on different occasions for different programmes) the board or boards of studies determine together the scheduling and frequency of re-examination occasions.

### Retakes of other forms of examination

Regulations concerning retakes of other forms of examination than written examinations and digital and computer-based examinations are given in the LiU guidelines for examinations and examiners, Dnr LiU-2023-00379

(<http://styrdokument.liu.se/Regelsamling/VisaBeslut/917592>).

In principle, other examination forms should be handled in the same way as a written examination when they are given for the last time. However, the times for the examination may vary based on the nature of the element compared to the times for the written examinations.

### Course closure

For Decision on Routines for Administration of the Discontinuation of Educational Programs, Freestanding Courses and Courses in Programs, see DNR LiU-2021-04782

(<https://styrdokument.liu.se/Regelsamling/VisaBeslut/1156410>). After a decision on closure and after the end of the discontinuation period, the students are referred to a replacement course (or similar) according to information in the course syllabus or programme syllabus. If a student has passed some part/parts of a closed program course but not all, and there is an at least partially replacing course, an assessment of crediting can be made. For questions about the crediting of course components, contact the Study counsellors.

### Registration for examination

In order to take a written, digital or computer-based examination, registration in advance is mandatory, see decision in the university's rule book Dnr LiU-2020-04559 (<https://styrdokument.liu.se/Regelsamling/VisaBeslut/622682>). An unregistered student can thus not be offered a place. The registration is done by the student at the Student Portal or in the LiU-app during the registration period. The registration period opens 30 days before the date of the examination and closes 10 days before the date of the examination. Candidates are informed of the location of the examination by email, four days in advance.

### Code of conduct for students during examinations

Details are given in a decision in the university's rule book, Dnr LiU-2020-04559 (<http://styrdokument.liu.se/Regelsamling/VisaBeslut/622682>).

### Retakes for higher grade

Students at the Faculty of Science and Engineering at LiU have the right to retake written examinations and digital and computer-based examinations in an attempt to achieve a higher grade. This is valid for all examination components with code "TEN", "DIT" and "DAT". The same right may not be exercised for other examination components, unless otherwise specified in the course syllabus.

A retake is not possible on courses that are included in an issued degree diploma.

### Grades

The grades that are preferably to be used are Fail (U), Pass (3), Pass not without distinction (4) and Pass with distinction (5).

- Grades U, 3, 4, 5 are to be awarded for courses that have written or digital examinations.

- Grades Fail (U) and Pass (G) may be awarded for courses with a large degree of practical components such as laboratory work, project work and group work.
- Grades Fail (U) and Pass (G) are to be used for degree projects and other independent work.

### Examination components

The following examination components and associated module codes are used at the Faculty of Science and Engineering:

- Grades U, 3, 4, 5 are to be awarded for written examinations (TEN) and digital examinations (DIT).
- Examination components for which the grades Fail (U) and Pass (G) may be awarded are laboratory work (LAB), project work (PRA), preparatory written examination (KTR), digital preparatory written examination (DIK), oral examination (MUN), computer-based examination in a computer lab (DAT), home assignment (HEM), assignment (UPG), and digital preparatory written examination in a computer lab (DAK).
- Students receive grades either Fail (U) or Pass (G) for other examination components in which the examination criteria are satisfied principally through active attendance such as tutorial group (BAS) or examination item (MOM).
- Grades Fail (U) and Pass (G) are to be used for the examination components Opposition (OPPO) and Attendance at thesis presentation (AUSK) (i.e. part of the degree project).

In general, the following applies:

- Mandatory course components must be scored and given a module code.
- Examination components that are not scored, cannot be mandatory. Hence, it is voluntary to participate in these examinations, and the voluntariness must be clearly stated. Additionally, if there are any associated conditions to the examination component, these must be clearly stated as well.
- For courses with more than one examination component with grades U,3,4,5, it shall be clearly stated how the final grade is weighted.

For mandatory components, the following applies (in accordance with the LiU Guidelines for education and examination for first-cycle and second-cycle education at Linköping University, Dnr LiU-2023-00379

<http://styrdokument.liu.se/Regelsamling/VisaBeslut/917592>):

- If special circumstances prevail, and if it is possible with consideration of the nature of the compulsory component, the examiner may decide to replace the compulsory component with another equivalent component.

For possibilities to alternative forms of examinations, the following applies (in accordance with the LiU Guidelines for education and examination for first-cycle and second-cycle education at Linköping University, Dnr LiU-2023-00379

<http://styrdokument.liu.se/Regelsamling/VisaBeslut/917592>):



- If the LiU coordinator for students with disabilities has granted a student the right to an adapted examination for a written examination in an examination hall, the student has the right to it.
- If the coordinator has recommended for the student an adapted examination or alternative form of examination, the examiner may grant this if the examiner assesses that it is possible, based on consideration of the course objectives.
- An examiner may also decide that an adapted examination or alternative form of examination if the examiner assessed that special circumstances prevail, and the examiner assesses that it is possible while maintaining the objectives of the course.

### **Reporting of examination results**

The examination results for a student are reported at the relevant department.

### **Plagiarism**

For examinations that involve the writing of reports, in cases in which it can be assumed that the student has had access to other sources (such as during project work, writing essays, etc.), the material submitted must be prepared in accordance with principles for acceptable practice when referring to sources when the text, images, ideas, data, etc. of other people are used. This is done by using references or quotations for which the source is specified. It is also to be made clear whether the author has reused his or her own text, images, ideas, data, etc. from previous examinations, such as degree projects, project reports, etc. (this is sometimes known as “self-plagiarism”).

A failure to specify such sources may be regarded as attempted deception during examination.

### **Use of generative AI**

Linköping University has produced a guide for teachers and students' use of generative AI in education (Dnr LiU-2023-02660). As a student, you are always expected to gain knowledge of what applies to each course (including the degree project). In general, clarity to where and how generative AI has been used is important.

### **Regulations (apply to LiU in its entirety)**

The university is a government agency whose operations are regulated by legislation and ordinances, which include the Higher Education Act and the Higher Education Ordinance. In addition to legislation and ordinances, operations are subject to several policy documents. The Linköping University rule book collects currently valid decisions of a regulatory nature taken by the university board, the vice-chancellor and faculty/department boards.

LiU's rule book for education at first-cycle and second-cycle levels is available at [http://styrdokument.liu.se/Regelsamling/Innehall/Utbildning\\_pa\\_grund-\\_och\\_avancerad\\_niva](http://styrdokument.liu.se/Regelsamling/Innehall/Utbildning_pa_grund-_och_avancerad_niva).

[Student rights and obligations \(login required\)](#)  
[Linköping University common rules and regulations \(in Swedish\)](#)