

# Surfaces and Interfaces

Programme course

6 credits

Ytor och gränsskikt

TFYA47

Valid from: 2019 Spring semester

**Determined by**

Board of Studies for Chemistry, Biology  
and Biotechnology

**Date determined**

2018-08-31

## Main field of study

Engineering Biology, Applied Physics, Physics

## Course level

Second cycle

## Advancement level

A1X

## Course offered for

- Engineering Biology, M Sc in Engineering
- Chemical Biology, M Sc in Engineering
- Biomedical Engineering, M Sc in Engineering

## Entry requirements

Note: Admission requirements for non-programme students usually also include admission requirements for the programme and threshold requirements for progression within the programme, or corresponding.

## Prerequisites

Basic knowledge of calculus in many variables, probability and statistics, classical physics, general chemistry, physical chemistry and quantum mechanics.

## Intended learning outcomes

The course gives a general introduction to surfaces; their structure and physical-chemical properties, and interfaces between solids and organic materials or liquids. This includes hard surfaces (crystalline and amorphous), soft matter surfaces (polymers and molecular films), free liquid surfaces and their interfaces, and colloidal systems. The course introduces analytical techniques for studies of surfaces, interfaces, thin films and colloids. After the course the student should be able to:

- Understand and describe properties of free liquid surfaces, such as surface tension, capillarity, wetting and spreading.
- Understand and describe electrical phenomena at surfaces, such as surface charge, surface potential, the electrical double layer, and basic electrochemical concepts.
- Describe the phase behaviour and aggregation of amphiphiles in solution and at interfaces.
- Describe intermolecular forces, forces acting between molecules and surfaces, and surface forces.
- Describe common crystal structures of solid materials and understand the arrangement of atoms in surfaces made from such crystals.
- Describe the mechanisms that control the formation of overlayer structures on single crystal surfaces and classify such overlayer structures.
- Describe the shape and properties of common adsorption isotherms, and apply them for extraction of thermodynamic or physicochemical data.
- Describe the function and principles for common methods and instruments for surface and thin film analysis, as well as explain and interpret the information that can be obtained from these techniques.
- Describe methods for the preparation and characterization of thin organic films on surfaces, in particular self-assembled monolayers.
- Describe and understand fundamental properties of colloidal systems, and methods for studies of such systems.

## Course content

Surface phases, surface energy, surface tension, surface excess. Free liquid surfaces, capillarity, Laplace' and Kelvin's equations. Wetting, contact angles, Young's equation. Surface charge, electrical double layer, Zeta potential. Amphiphilic aggregation and phase behaviour. Adsorption to liquid interfaces, Gibb's isotherm. Forces between molecules and between molecular systems and surfaces. Colloids and colloidal stability. The structure and properties of molecular films; preparation strategies, thermodynamically and kinetically controlled processes, surface modification, self-assembled monolayers. Hard materials and surfaces, the structure of surfaces, the surface chemical bond, physisorption, chemisorption. Adsorption, adsorption kinetics and isotherms, surface reactions. Analytical methods for analysis of surfaces, interfaces and colloids.

## Teaching and working methods

The material is presented in lectures, as well as classroom and laboratory exercises.

## Examination

LAB1	Laboratory course	1.5 credits	U, G
TEN2	Written examination	4.5 credits	U, 3, 4, 5

## Grades

Four-grade scale, LiU, U, 3, 4, 5

## Other information

Supplementary courses: Surface science, Biosensor technology, Microsystems and Nanobiology, Biomedical materials.

## Department

Institutionen för fysik, kemi och biologi

## Director of Studies or equivalent

Magnus Boman

## Examiner

Thomas Ederth

## Course website and other links

<https://www.ifm.liu.se/edu/coursescms/tfya47/>

## Education components

Preliminary scheduled hours: 48 h

Recommended self-study hours: 112 h

## Course literature

### Books

Attard, Gary, Barnes, Colin, (1998) *Surfaces* Oxford : Oxford University Press, 1998

ISBN: 0198556861, 9780198556862

Barnes, Geoffrey, Gentle, Ian, (2011) *Interfacial science : an introduction* 2nd ed.

Oxford : Oxford University Press, c2011

ISBN: 9780199571185, 019957118x

### Other

#### Course literature

The literature consists of the two books listed below, and additional material on electrochemistry and cyclic voltammetry, which is available from the course webpage under "Lab exercises".

- G.T. Barnes & I.R. Gentle, *Interfacial science: An introduction*, 2nd ed., Oxford University Press 2011, ISBN 9780199571185.
- G. Attard & C. Barnes, *Surfaces*, Oxford University Press 1998, ISBN 9780198556862.

#### Other literature

As far as I am aware, there are no single books covering the whole course contents, but here is one which almost does, and which could be good alternative reading:

- H.-J. Butt, K. Graf, M. Kappl, *Physics and Chemistry of Interfaces*, 3 uppl., Wiley-VCH 2013, ISBN 9783527412167.

Other literature for further reading:

- J.C. Berg, *An Introduction to Interfaces and Colloids*, World Scientific 2009, ISBN 9789814299824 (emphasis on surface and colloid chemistry, liquid interfaces).
- J.N. Israelachvili, *Intermolecular and surface forces*, 3 uppl., Academic Press, 2011, ISBN 9780123751829 (stand literature on intermolecular and surface forces).
- A.W. Adamson and A.P. Gast, *Physical chemistry of surfaces*, 6 uppl., Wiley 1997, ISBN 0471148733 (A classic in the field, with both fundamentals and applications).
- H.Y. Erbil, *Surface Chemistry of Solid and Liquid Interfaces*, John Wiley & Sons 2006, ISBN 9781405119689 (some emphasis on thermodynamic aspects).

## Common rules

### Course syllabus

A syllabus has been established for each course. The syllabus specifies the aim and contents of the course, and the prior knowledge that a student must have in order to be able to benefit from the course.

### Timetabling

Courses are timetabled after a decision has been made for this course concerning its assignment to a timetable module. A central timetable is not drawn up for courses with fewer than five participants. Most project courses do not have a central timetable.

### Interrupting a course

The vice-chancellor's decision concerning regulations for registration, deregistration and reporting results (Dnr LiU-2015-01241) states that interruptions in study are to be recorded in Ladok. Thus, all students who do not participate in a course for which they have registered must record the interruption, such that the registration on the course can be removed. Deregistration from a course is carried out using a web-based form: [www.lith.liu.se/for-studenter/kurskomplettering?l=sv](http://www.lith.liu.se/for-studenter/kurskomplettering?l=sv).

### Cancelled courses

Courses with few participants (fewer than 10) may be cancelled or organised in a manner that differs from that stated in the course syllabus. The board of studies is to deliberate and decide whether a course is to be cancelled or changed from the course syllabus.

### Regulations relating to examinations and examiners

Details are given in a decision in the university's rule book:  
<http://styrdokument.liu.se/Regelsamling/VisaBeslut/622678>.

### Forms of examination

#### Examination

Written and oral examinations are held at least three times a year: once immediately after the end of the course, once in August, and once (usually) in one of the re-examination periods. Examinations held at other times are to follow a decision of the board of studies.

Principles for examination scheduling for courses that follow the study periods:

- courses given in VT1 are examined for the first time in March, with re-

examination in June and August

- courses given in VT2 are examined for the first time in May, with re-examination in August and October
- courses given in HT1 are examined for the first time in October, with re-examination in January and August
- courses given in HT2 are examined for the first time in January, with re-examination at Easter and in August.

The examination schedule is based on the structure of timetable modules, but there may be deviations from this, mainly in the case of courses that are studied and examined for several programmes and in lower grades (i.e. 1 and 2).

- Examinations for courses that the board of studies has decided are to be held in alternate years are held only three times during the year in which the course is given.
- Examinations for courses that are cancelled or rescheduled such that they are not given in one or several years are held three times during the year that immediately follows the course, with examination scheduling that corresponds to the scheduling that was in force before the course was cancelled or rescheduled.
- If teaching is no longer given for a course, three examination occurrences are held during the immediately subsequent year, while examinations are at the same time held for any replacement course that is given, or alternatively in association with other re-examination opportunities. Furthermore, an examination is held on one further occasion during the next subsequent year, unless the board of studies determines otherwise.
- If a course is given during several periods of the year (for programmes, or on different occasions for different programmes) the board or boards of studies determine together the scheduling and frequency of re-examination occasions.

### **Registration for examination**

In order to take an examination, a student must register in advance at the Student Portal during the registration period, which opens 30 days before the date of the examination and closes 10 days before it. Candidates are informed of the location of the examination by email, four days in advance. Students who have not registered for an examination run the risk of being refused admittance to the examination, if space is not available.

Symbols used in the examination registration system:

\*\* denotes that the examination is being given for the penultimate time.

\* denotes that the examination is being given for the last time.

### **Code of conduct for students during examinations**

Details are given in a decision in the university's rule book:  
<http://styrdokument.liu.se/Regelsamling/VisaBeslut/622682>.

### **Retakes for higher grade**

Students at the Institute of Technology at LiU have the right to retake written examinations and computer-based examinations in an attempt to achieve a higher grade. This is valid for all examination components with code "TEN" and "DAT". The same right may not be exercised for other examination components, unless otherwise specified in the course syllabus.

### **Retakes of other forms of examination**

Regulations concerning retakes of other forms of examination than written examinations and computer-based examinations are given in the LiU regulations for examinations and examiners,

<http://stydokument.liu.se/Regelsamling/VisaBeslut/622678>.

### **Plagiarism**

For examinations that involve the writing of reports, in cases in which it can be assumed that the student has had access to other sources (such as during project work, writing essays, etc.), the material submitted must be prepared in accordance with principles for acceptable practice when referring to sources (references or quotations for which the source is specified) when the text, images, ideas, data, etc. of other people are used. It is also to be made clear whether the author has reused his or her own text, images, ideas, data, etc. from previous examinations.

A failure to specify such sources may be regarded as attempted deception during examination.

### **Attempts to cheat**

In the event of a suspected attempt by a student to cheat during an examination, or when study performance is to be assessed as specified in Chapter 10 of the Higher Education Ordinance, the examiner is to report this to the disciplinary board of the university. Possible consequences for the student are suspension from study and a formal warning. More information is available at <https://www.student.liu.se/studenttjanster/lagar-regler-rattigheter?l=sv>.

### **Grades**

The grades that are preferably to be used are Fail (U), Pass (3), Pass not with distinction (4) and Pass with distinction (5). Courses under the auspices of the faculty board of the Faculty of Science and Engineering (Institute of Technology) are to be given special attention in this regard.

1. Grades U, 3, 4, 5 are to be awarded for courses that have written examinations.
2. Grades Fail (U) and Pass (G) may be awarded for courses with a large degree of practical components such as laboratory work, project work and group work.

### **Examination components**

1. Grades U, 3, 4, 5 are to be awarded for written examinations (TEN).
2. Grades Fail (U) and Pass (G) are to be used for undergraduate projects and other independent work.



3. Examination components for which the grades Fail (U) and Pass (G) may be awarded are laboratory work (LAB), project work (PRA), preparatory written examination (KTR), oral examination (MUN), computer-based examination (DAT), home assignment (HEM), and assignment (UPG).
4. Students receive grades either Fail (U) or Pass (G) for other examination components in which the examination criteria are satisfied principally through active attendance such as other examination (ANN), tutorial group (BAS) or examination item (MOM).

The examination results for a student are reported at the relevant department.

### **Regulations (apply to LiU in its entirety)**

The university is a government agency whose operations are regulated by legislation and ordinances, which include the Higher Education Act and the Higher Education Ordinance. In addition to legislation and ordinances, operations are subject to several policy documents. The Linköping University rule book collects currently valid decisions of a regulatory nature taken by the university board, the vice-chancellor and faculty/department boards.

LiU's rule book for education at first-cycle and second-cycle levels is available at [http://styrdokument.liu.se/Regelsamling/Innehall/Utbildning\\_pa\\_grund-\\_och\\_avancerad\\_niva](http://styrdokument.liu.se/Regelsamling/Innehall/Utbildning_pa_grund-_och_avancerad_niva).